

STEREOCHEMISTRY OF ELECTROPHILIC ADDITION TO DIMETHYL TRICYCLO[4,2,2,0^{2,5}]DECA-3,7-DIENE-9,10-DICARBOXYLATE. MOLECULAR STRUCTURE OF Hg(OAc)₂-ADDITION PRODUCT

Nikolai S.Zefirov*, A.S.Koz'min and V.N.Kirin

Department of Chemistry, Moscow State University, Moscow 117234, USSR

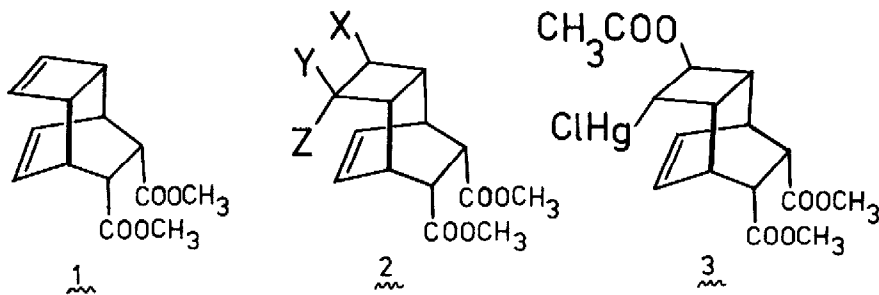
B.B.Sedov and V.G.Rau

Vladimir Pedagogical Institute, Vladimir, USSR

Summary - The molecular structure of the trans addition product of Hg(OAc)₂ to diester 1 has been determined by X-ray diffraction (represented by formula 3) which leads to a revision of previously claimed cis-additions in this series.

Commonly accepted suggestions about the formation of the bridged intermediates (onium ion¹, π -complex^{1b}, ion-pairs² etc.) in Ad_E-reactions¹ are based mainly on the observed trans-stereochemistry of these processes.

Hence, rare cases of cis addition always attract the attention. Some particular types of olefine structure are believed to be especially liable to undergo unusual cis additions, and the diester 1 is one of them. For example, T.Sasaki and co-workers³ and G.Mehta and P.N.Pandey^{4a} have stated that oxymercuration of diester 1 occurs on cyclobutene moiety as cis addition to yield the corresponding cis-oxymercurials 2 (X=OR, Y=HgCl, Z=H). Similar reaction course has been reported^{4b} for the addition of iodine azide to diester 1.



Both adducts $\underline{2}$ ($X=OR$, $Y=HgCl$, $Z=H$) and $\underline{2}$ ($X=I$, $Y=N_3$, $Z=H$) were shown to have similar stereochemistry by chemical correlation and their cis-configuration has been deduced from 1H NMR data.

This alleged exclusive cis addition has been interpreted in term of twist strain theory. Additionally, G.Mehta and P.N.Pandey reported "the first non-stereospecific addition" of $PhSCl$ to diester $\underline{1}$ which yields the mixture of trans adduct, $\underline{2}$ ($X=PhS$, $Y=H$, $Z=Cl$) and cis adduct, $\underline{2}$ ($X=PhS$, $Y=Cl$, $Z=H$)⁵ (cf. ⁶).

However, the thorough analysis of the experimental data for these "abnormal stereochemical results" reveals some discrepancies, which necessitates the revision of the whole picture.

Firstly, it is well known that stereochemistry of addition of mercuric salts depends on the structure of olefins ⁷. However, the cyclobutene undergoes trans addition ⁸ and hence the data of Refs ^{3,4} cannot be correlated with the wealth of the other data on the structural effects governing the stereochemistry of oxymercuration reactions. Moreover, iodine azide is known to give trans adducts ⁹ exclusively and this fact makes doubtful cis-structure ascribed, $\underline{2}$ ($X=I$, $Y=N_3$, $Z=H$)^{4b}.

Secondly, the cis addition of $Hg(OAc)_2$ in methanol proceeds usually giving the mixture of acetoxy and methoxy mercurials while the formation of the single methoxy mercurial is typical result for the trans addition process⁷. In contrast, the oxymercuration of diester $\underline{1}$ with $Hg(OAc)_2$ in methanol proceeds surprisingly without formation of acetoxy mercuration product ^{3,4a}.

Thirdly, careful examination of the NMR criteria used for the configurational assignment of the compounds of type $\underline{2}$ revealed some severe inconsistencies in this approach ^{6,10}. These considerations allowed us to suggest that aforementioned Ad_E -reactions to $\underline{1}$ actually proceed as trans addition and the conclusions in Refs. ³⁻⁵ are based on the misinterpretation of the 1H NMR spectra.

Previously we attempted to reproduce the results of Ref. ⁵, but invariably obtained only trans adducts $\underline{2}$ ($X=ArS$, $Y=H$, $Z=Cl$, Br) ⁶ in reactions of $ArSZ$ with diester $\underline{1}$; their structures have been unambiguously proved by X-ray method ¹⁰.

We have repeated the reaction of $\text{Hg}(\text{OAc})_2$ with diester 1 in AcOH 3a,4a and investigated the structure of the adduct (m.p. 182° from ethylacetate) 12 by X-ray techniques. Experimental X-ray data were recorded using Syntex P2₁ autodiffractometer, Mo $K\alpha$. Crystals of the adduct are monoclinic, P2₁/n. The cell dimensions are $a=10.273(6)$, $b=15.316(8)$, $c=12.226(6)$ Å, $\beta=112.86^\circ$, $D_{\text{calc.}}=2.032$. The structure was solved by Patterson method and refined to $R_{\text{hkl}}=0.078$ for 2272 independent reflections. The X-ray molecular structure of the adduct is shown in Fig. 1 and represented by formula 3. There are two

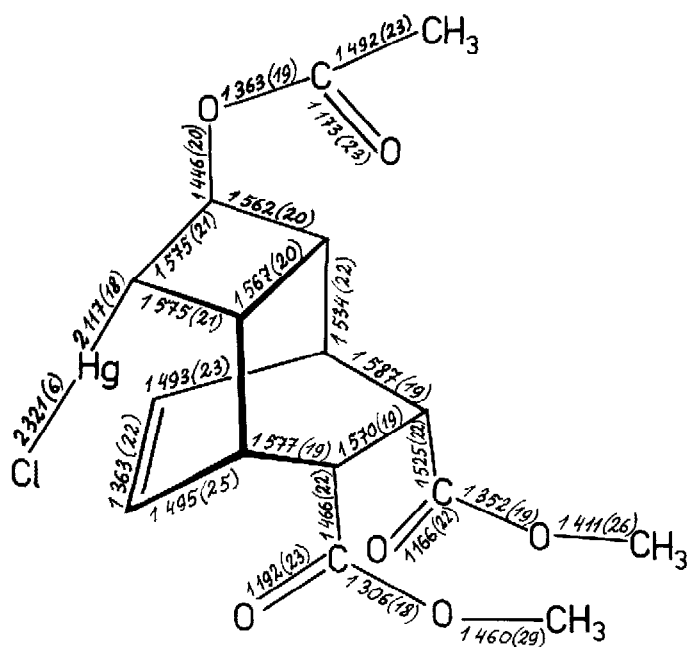


Fig. 1. molecular structure of dimethyl trans-3-endo-chloromercur-4-acetoxy-tricyclo[4,2,2,0^{2,5}]dec-7-ene-9,10-cis-endo-dicarboxylate.

points which are clearly evident from our X-ray results: 1) oxymercuration of the diester 1 occurs as trans addition in contrast to the suggestions made in Refs. 3a,4a. Hence all related cis structures must be revised or reinvestigated; 2) the initial attack of the cyclobutene double bond by electrophilic mercury salt is directed from endo-side. It is noteworthy that addition of other electrophiles such as halogenes 13, sulfenyl halides 6,13 etc. had been shown to proceed as an attack on the exo-side. Such an unusual mode of the initial

approach of an attacking electrophile is probably due to the coordination of mercury ion by two double bonds (the discussion of this problem see 7,14). This type of an attack on diester 1 has been also observed in oxypalladation reaction 3b.

REFERENCES AND NOTES.

1. (a) I.Roberts and G.E.Kimball, J.Am.Chem.Soc., 59, 947 (1937);
(b) M.J.S.Dewar and J.R.Ford, J.Am.Chem.Soc., 101, 783 (1979).
2. W.A.Smit, N.S.Zefirov, I.V.Bodrikov and M.Z.Krimer, Acc.Chem.Res., 12, 282 (1979).
3. (a) T.Sasaki, K.Kanematsu, A.Kondo and Y.Nishitani, J.Org.Chem., 39, 3569 (1974); (b) T.Sasaki, K.Kanematsu and A.Kondo, J.Chem.Soc., Perkin 1, 2516 (1976).
4. (a) G.Mehta and P.N.Pandey, J.Org.Chem., 40, 3631 (1975);
(b) G.Mehta, P.K.Dutta and P.N.Pandey, Tetrahedron Letters, 445 (1975).
5. G.Mehta and P.N.Pandey, Tetrahedron Letters, 3567 (1975).
6. N.S.Zefirov, V.N.Kirin, K.A.Potekhin, A.S.Koz'min, N.K.Sadovaya, E.N.Kurkutova and I.V.Bodrikov, Zhur.Org.Khim., 14, 1224 (1978).
7. N.S.Zefirov, Usp.Khim., 34, 1272 (1965).
8. T.G.Traylor, Acc.Chem.Res., 2, 152 (1969); J.L.Waters, Tetrahedron Letters, 3770 (1969).
9. E.W.Fowler, A.Hassner and L.A.Levy, J.Am.Chem.Soc., 89, 2077 (1967).
10. K.A.Potekhin, E.N.Kurkutova, Yu.T.Struchkov, M.Yu.Antipin, V.N.Kirin, A.S.Koz'min and N.S.Zefirov, Dokl.Akad.Nauk SSSR, 242, 341 (1978);
K.A.Potekhin, E.N.Kurkutova, Yu.T.Struchkov, A.S.Koz'min, V.N.Kirin, N.S.Zefirov, V.V.Ilyukhin and N.V.Belov, Ibid, 242, 832 (1978);
K.A.Potekhin, E.N.Kurkutova, Yu.T.Struchkov, V.N.Kirin, A.S.Koz'min, N.S.Zefirov, V.V.Ilyukhin and N.V.Belov, Ibid, 243, 652 (1978).
11. J.P.Snyder and D.G.Farnum, J.Org.Chem., 31, 1699 (1976).
12. The reaction mixture ^{3,4a,6} was poured (instead of evaporating, cf. ^{3a,4a}) into excess of brine, the solid was filtered, washed with water, dried in vacuum and recrystallised from AcOEt (recrystallisation from methanol ^{3a,4a} yielded the product with inorganic impurities) to give 89% of 3.
13. A.Kondo, T.Yamane, T.Ashida, T.Sasaki and K.Kanematsu, J.Org.Chem., 43, 1180 (1978).
14. N.S.Zefirov and L.G.Gurvich, J.Organomet.Chem., 81, 309 (1974);
N.Takaishi, Y.Fujikura, Y.Inamoto, J.Org.Chem., 40, 3767 (1975).

(Received in UK 28 November 1979)